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part 1 ( Chem 1 H)**

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"The Three States of  
Matter" The Creation

of Chemistry - The  
Fundamental Laws:

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1—Introduction:

Matter and

Measurement States  
of matter | States of  
matter and

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*intermolecular forces /*

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*Change Part 1*

*Particulate nature of*

*matter - Part-1 -*

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Chemistry **Changes  
of States of Matter**  
Change Chapter  
13 **Answers**  
**Physical Changes**

---

States of Matter :

Solid Liquid Gas 10

*Amazing Experiments  
with Water*

---

Pure Substances and  
Mixtures!

(Classification of  
Matter) Physical and  
Chemical Changes

---

Understanding Atoms,  
*Page 7/37*

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elements, and  
molecules Part #1  
(9min)

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Basic Chemistry  
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matter? - HD  
Properties of Matter  
Physical and  
Chemical Changes:  
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Changes States of  
Matter - Solids,  
Liquids, Gases

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Chemistry Change of  
State | Matter |  
Physics | Fuse School~~  
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Changes of State -  
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\u0026 Changing  
State #20 Pure  
Substances and  
Mixtures, Elements  
\u0026 Compounds,

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Classification of  
Matter, Chemistry  
Examples,

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8th/Chapter 1 /Matter

~~Classifying Matter  
With Practice~~

~~Problems | Study  
Chemistry With Us~~

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Stories of Two  
Chemicals Ozone  
Layer, atmosphere,  
ozone formation,  
chlorofluorocarbons,  
CFC's 1.2 Chemistry  
and Matter Chemistry  
Central Science

## **Chapter Summaries – Chemistry Matter and Change**

Chapter 1 Section 2

*Page 12/37*

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MATTER and Its  
Properties Matter –  
Anything that has  
MASS and VOLUME.  
ATOMS and  
MOLECULES the  
basic building blocks  
of matter. ATOM-- the  
smallest unit of an  
element that still  
maintains the  
properties that  
element. ELEMENT–  
a pure substance

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Matter and Change Chapter  
made up of only one  
type of atom.

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**MATTER and  
CHANGE Chapter 1  
Section 1**

Chemistry: Matter and  
Change Chapter 6  
Study Guide for  
Content Mastery .

Name Date

CHAPTER Class  
STUDY GUIDE FOR  
CONTENT

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6.3 Periodic Trends In  
your textbook, read  
about atomic radius  
and ionic radius.

Circle the letter of the  
choice that best  
completes the  
statement or answers  
the question. 1.

Atomic radii cannot be  
measured ...

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CHAPTER 3

SOLUTIONS

MANUAL

Matter—Properties and

ChangesMatter—Prop

erties and Changes

Solutions Manual

Chemistry: Matter and

Change • Chapter 3

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## 35 Section 3.1

Properties of Matter  
pages 70–75 Problem-  
Solving Lab 1. Explain  
why the flow of a  
compressed gas must  
be controlled for  
practical and safe  
use.

## **Matter—Properties and Changes**

### **Matter— Properties and Changes**

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Chapter 6 – The  
Periodic Table and  
Periodic Law Section

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6.1 Development of  
the Modern Periodic  
Table 1. octaves 2.  
eight 3. nine 4.  
accepted 5. Dmitri  
Mendeleev 6. atomic  
mass 7. elements 8.  
atomic number 9.  
Henry Moseley 10.  
protons 11. periodic  
law 12. properties 13.  
14 ...

## **Ch 6 Study Guide**

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## Answers And

Date Class Chapter  
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Assessment

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Chemistry: Matter and  
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25 Electrons in Atoms  
Reviewing Vocabulary

Match the definition in  
Column A with the  
term in Column B.

Column A Column B

1. The set of  
frequencies of the  
electromagnetic

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waves emitted by the atoms of an element  
2. The minimum amount of energy that can be lost or gained by ...

## **Chapter Assessment**

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incorporates features  
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Matter and problem-solving development.

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some numbers used

in chemistry can be

q... the study of energy

changes (that is,

energy produced or

absor... A system that

can exchange neither

energy nor matter

with its... Chemistry

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extensive property.  
intensive property.  
extensive properties  
(examples) intensive  
property (examples) a  
property that depends  
on the amount of  
matter in a sample. a

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property that depends  
on the type of matter  
in a sample not....

mass, volume, length,  
weight.

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Math Skills

Transparency

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## Chapter 11 31

number of moles of  
known substance C H

C H C H O O O CO

CO CO H O H O H O

number of moles of  
unknown substance O

CO H O C H CO H O

C H O H O C H O CO

use mole ratio 5 mol

O 2 1 mol C 3 H 8 3

mol CO 2 1 mol C 3 H

8 4 mol H 2 O 1 mol C

3 H 8 1 mol C 3 H 8 5

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**MATH SKILLS**  
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A “chemical change”  
is a change that  
produces matter with  
a different  
composition than the  
original matter.

Chemical Change A  
change in which one  
or more substances

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are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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