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Pre-laboratory Work Before the experiment in the laboratory, you should be able to answer these questions. 1) List five physical properties. 2) What are the differences between physical and chemical changes? 3) Define solubility and discuss the factors affecting solubility. 4) What is density? How do you calculate it? CHEM 102 GENERAL CHEMISTRY

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Mass or weight 3Gram (g) = weight of 1 cubic centimeter (cm) of water at 4°C and 760 mm Hg Mass = quantity of material Weight = mass x gravitational force Length Meter (m) =100 cm =1000 millimeters (mm) =39.37 in. Volume Liter (L) =volume of 1 kilogram (kg) of H. 2.

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General Chemistry Lab Manual Answer This manual, Introduction to Chemistry Lab Manual, is written for undergraduate students taking a General Chemistry laboratory course concurrently with General Chemistry. This is written primarily for those taking Chem 171 at Rutgers, The State University of New Jersey, however would be

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