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Overpotential:charts the twists and turns in the ongoing quest to create the perfect fuel cell. By exploring the gap between the theory and practice of fuel cell power, Matthew N. Eisler opens a window into broader issues in the history of science, technology, and society after the Second World War, including the sociology of laboratory life, the relationship between academe, industry, and government in developing advanced technologies, the role of technology in environmental and pollution ...

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~~Wednesday's Book Review: "Overpotential" by Roger Launius's~~

In electrochemistry, overpotential is the potential difference between a half-reaction's thermodynamically determined reduction potential and the potential at which the redox event is experimentally observed. The term is directly related to a cell's voltage efficiency. In an electrolytic cell the existence of overpotential implies the cell requires more energy than thermodynamically expected to drive a reaction. In a galvanic cell the existence of overpotential means less energy is recovered the

~~Overpotential - Wikipedia~~

There have been successes and serendipitous discoveries along the way, but engineering a fuel cell that is both durable and affordable has proved extraordinarily difficult.Overpotential charts the twists and turns in the ongoing quest to create the perfect fuel cell.

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